

## Stability of Functionalized Carbon Nanotubes

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Functionalization of carbon nanotubes (CNTs) with simple organic molecules plays an important role in applications of CNTs, either as electrical sensors of chemical and/or biological substances or as an enforcement of weak materials (like polymers) to obtain strong composite materials. Recently it has been observed a lot of research activity in these fields. We have performed extensive studies to determine the stability of the functionalized CNTs with simple organic molecules such as  $\text{CH}_n$ ,  $\text{NH}_n$ ,  $\text{OH}$ , and  $\text{COOH}$ . We discuss the dependence of the cohesive energies of functionalized CNTs, its deformations, and changes in the electronic structure on the density of the adsorbed molecules. All these factors influence the electrical and mechanical properties of the functionalized CNTs and are important for reliable description of the sensor electrical characteristics and also for modeling composite materials. Our studies are based on the *ab initio* calculations in the framework of the density functional theory. We use the generalized gradient approximation (GGA) for the exchange-correlation density functional and supercell geometry within the numerical package SIESTA.

Our studies reveal physical mechanisms of the binding of the studied groups to the CNTs. Generally, the stability of the functionalized CNTs is weakly dependent on the diameter of the CNTs, whereas the stability strongly decreases with the density of the adsorbed groups. In particular, we find that the  $\text{NH}_n$  and  $\text{CH}_n$  groups with  $n$  larger than two do not bind to the CNTs. In particular, it turns out that practically only  $\text{CH}_2$  groups make reasonably strong bonds to the CNTs ( $\text{CH}_3$  groups bind, but extremely weakly with bond length of 1.56 Å). In the case of  $\text{CH}_4$  groups, we observe their dissociation into  $\text{CH}_2$  and  $\text{H}_2$  dimer placed in the surrounding of CNT;  $\text{CH}_2$  binds to the CNTs, whereas  $\text{H}_2$  remains unbound. The functionalization of CNTs with  $\text{NH}_n$  causes practically no deformation of the CNTs, whereas the  $\text{CH}_n$  groups attached to the CNTs cause reconstruction of the CNTs.